



Measuring particulate organic and elemental carbon: challenges and proposed solutions

Jean-Philippe Putaud

European Commission, Joint Research Centre,
Institute for Environment and Sustainability, Air and Climate Unit



**DIRECTIVE 2008/50/EC OF THE EUROPEAN PARLIAMENT AND OF THE COUNCIL
of 21 May 2008
on ambient air quality and cleaner air for Europe**

**CHAPTER II
ASSESSMENT OF AMBIENT AIR QUALITY
SECTION 1**

***Assessment of ambient air quality in relation to sulphur
dioxide, nitrogen dioxide and oxides of nitrogen, particulate
matter, lead, benzene and carbon monoxide***

Article 5

Assessment regime

5. In addition to the assessments referred to in paragraphs 2, 3 and 4, **measurements shall be made, at rural background locations** away from significant sources of air pollution, for the purposes of providing, as a minimum, information on the total mass concentration and **the chemical speciation concentrations of fine particulate matter (PM_{2,5})** on an annual average basis ...

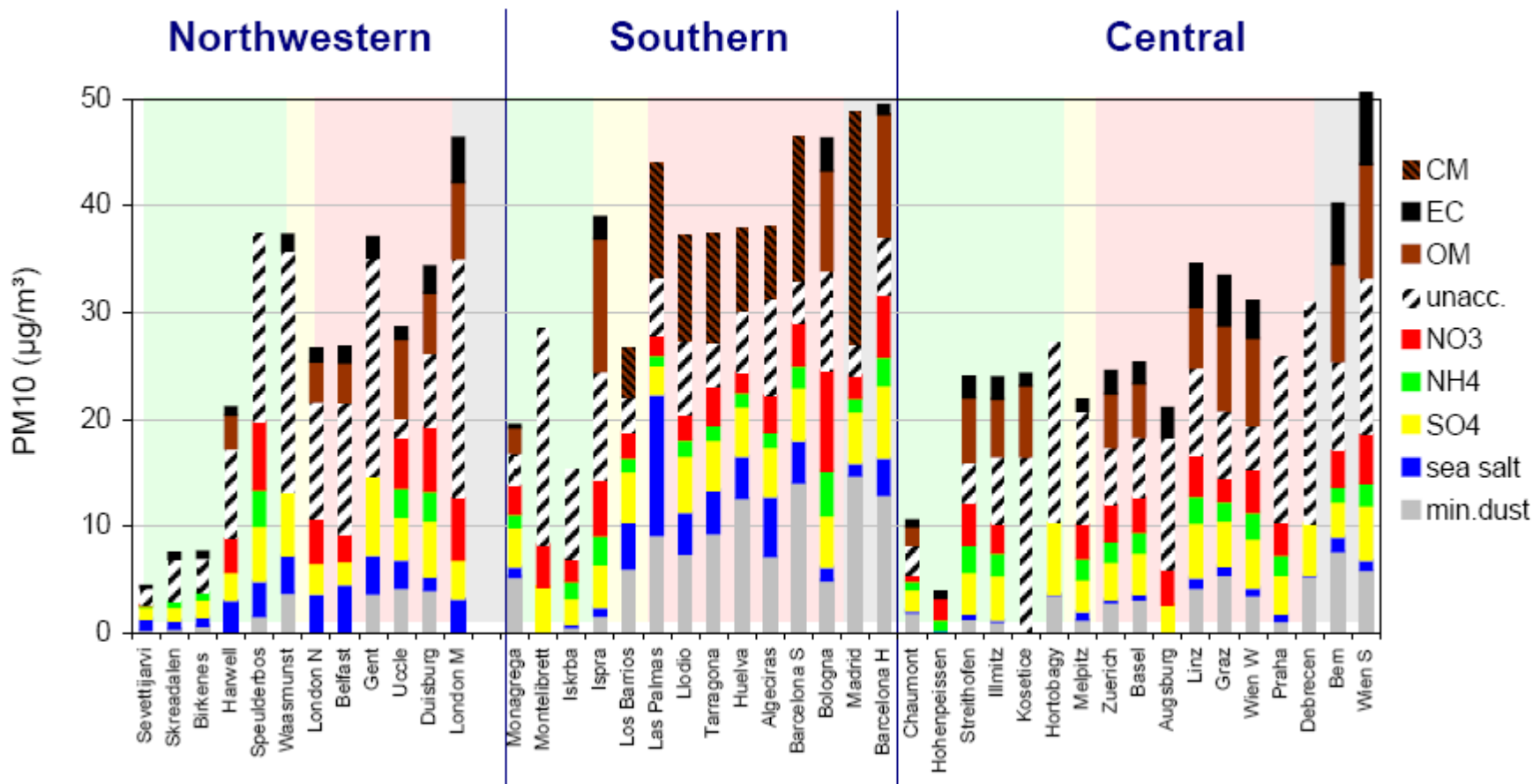
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Annex IV: At least the list of chemical species given below shall be included.

SO_4^{2-}	Na^+	NH_4^+	Ca^{2+}	elemental carbon (EC)
NO_3^-	K^+	Cl^-	Mg^{2+}	organic carbon (OC)

*The general term “soot” refers to combustion-generated aerosol mixtures of **two components**: first, the most thermally refractory and light absorbing byproduct of incomplete combustion, which is commonly referred to as elemental carbon (**EC**; thermal/optical analysis) or black carbon (BC; light absorption methods); and second, organic carbon (**OC**) that can have a wide range of thermal and light-absorbing properties (Baumgardner et al., 2012).*

Chemical characteristics of particulate matter from 43 rural, urban, and kerbside sites across Europe



Putaud et al., 2010

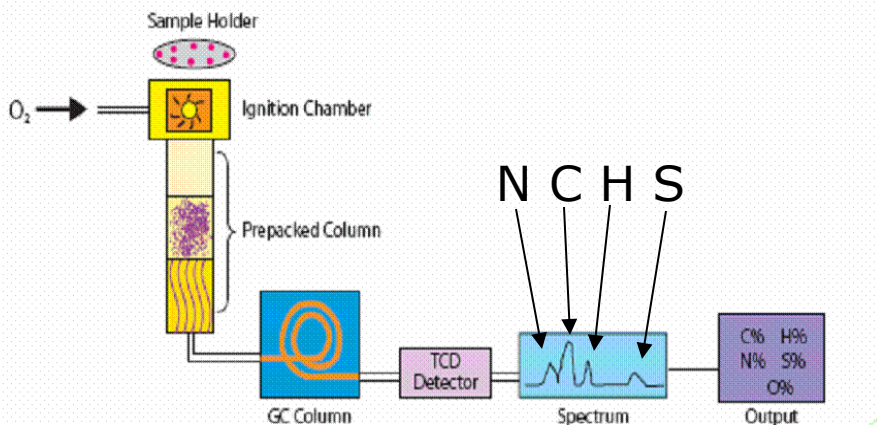
- Carbonaceous species account for a substantial fraction of PM_{2.5} (& PM₁₀)
- There were many sites where carbonaceous species were not measured
- The lack of reference methods prevents these measurements from becoming more “popular”
 - **sampling**
 - **analyses**
- What challenges should a practicable method for the analysis of particulate carbonaceous matter overcome ?

Carbonaceous atmospheric particulate matter consists of 1000's of species **with very different properties**

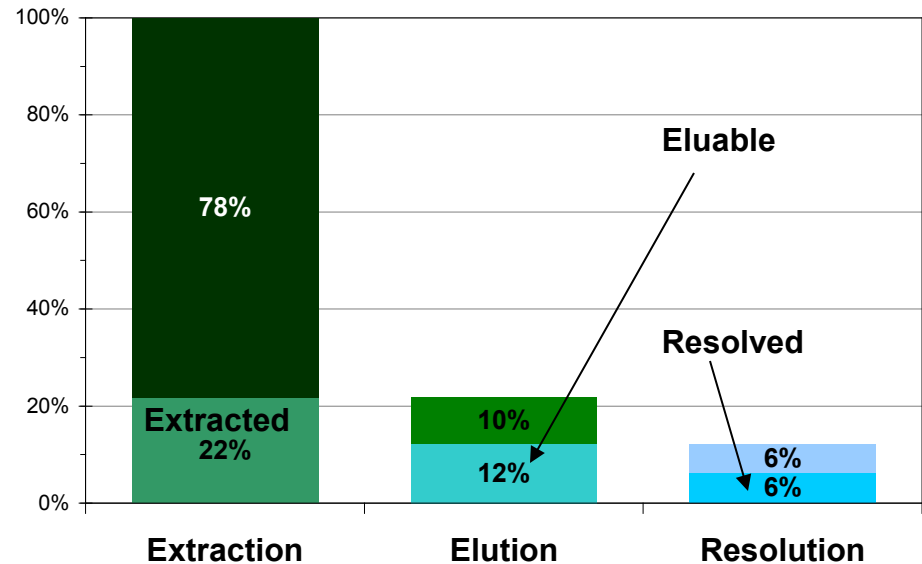
Atmospheric particulate matter speciation → resolve and quantify up to 10-15% of TC only

Elemental Analysis (CHN)

→ get Total Carbon (TC) amount

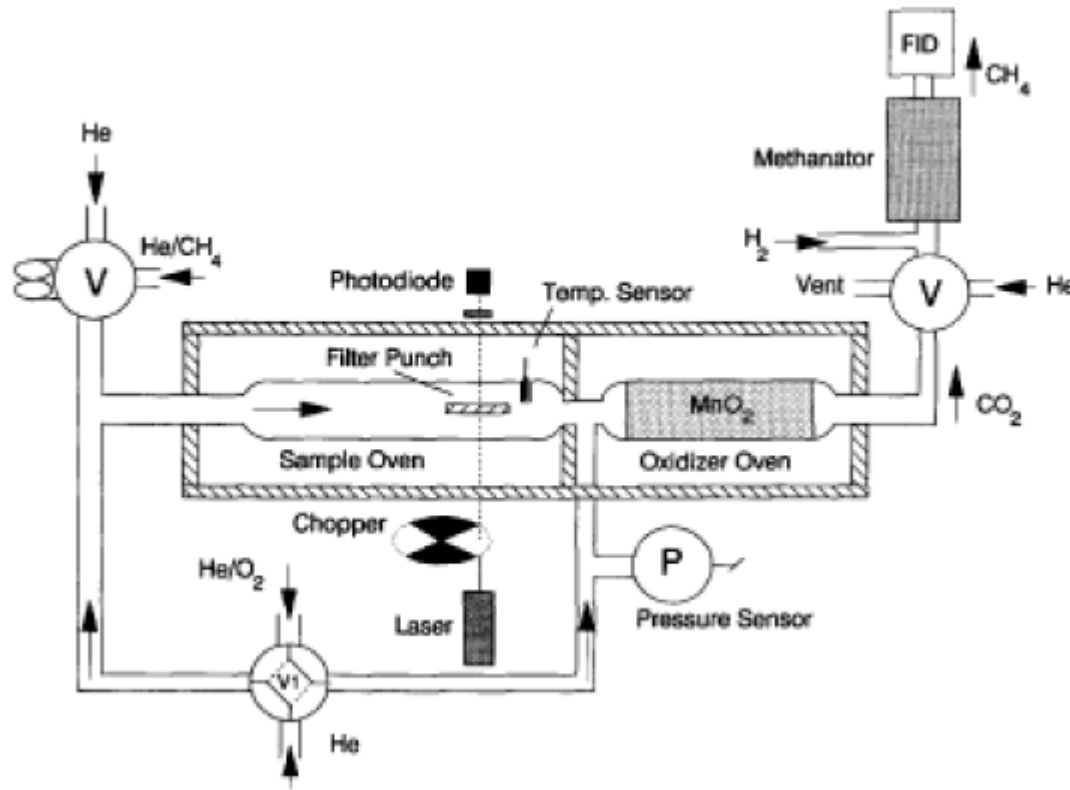


Carbonaceous matter speciation at Kuala Lumpur

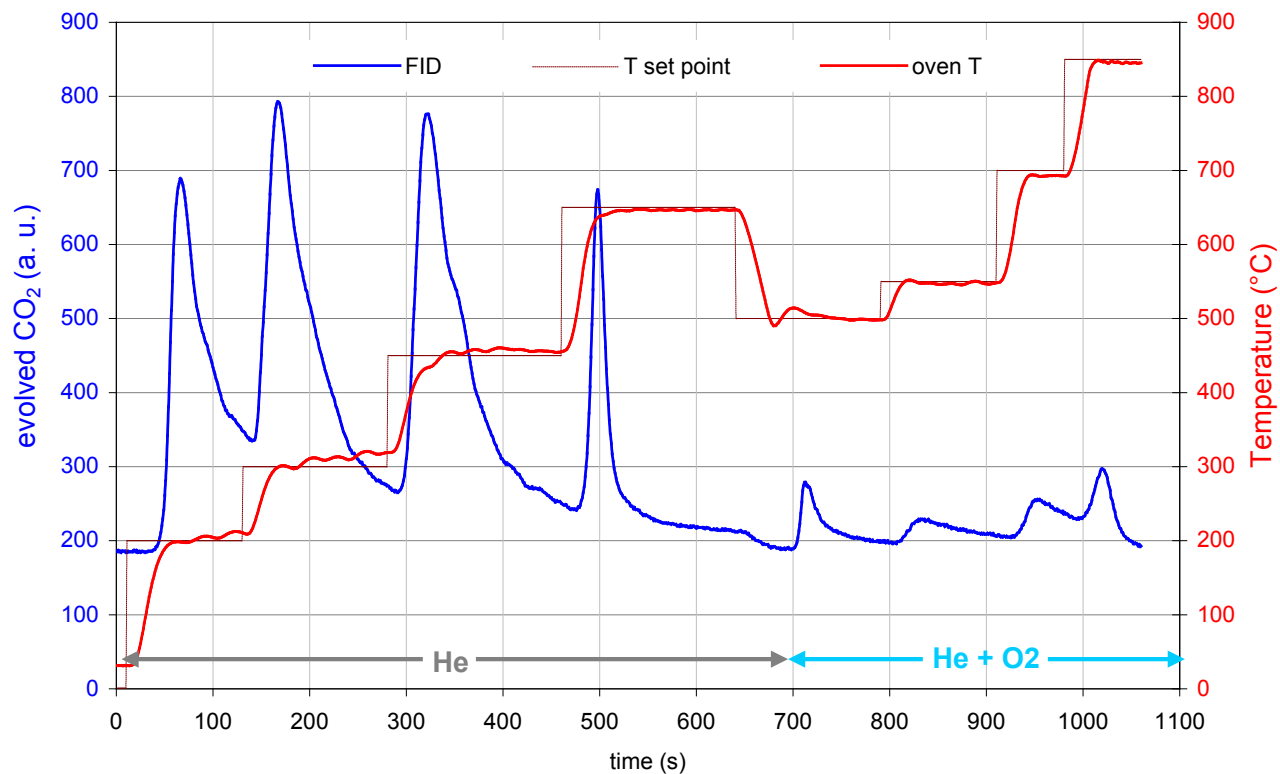


Adapted from Alam et al. (2004) and references therein

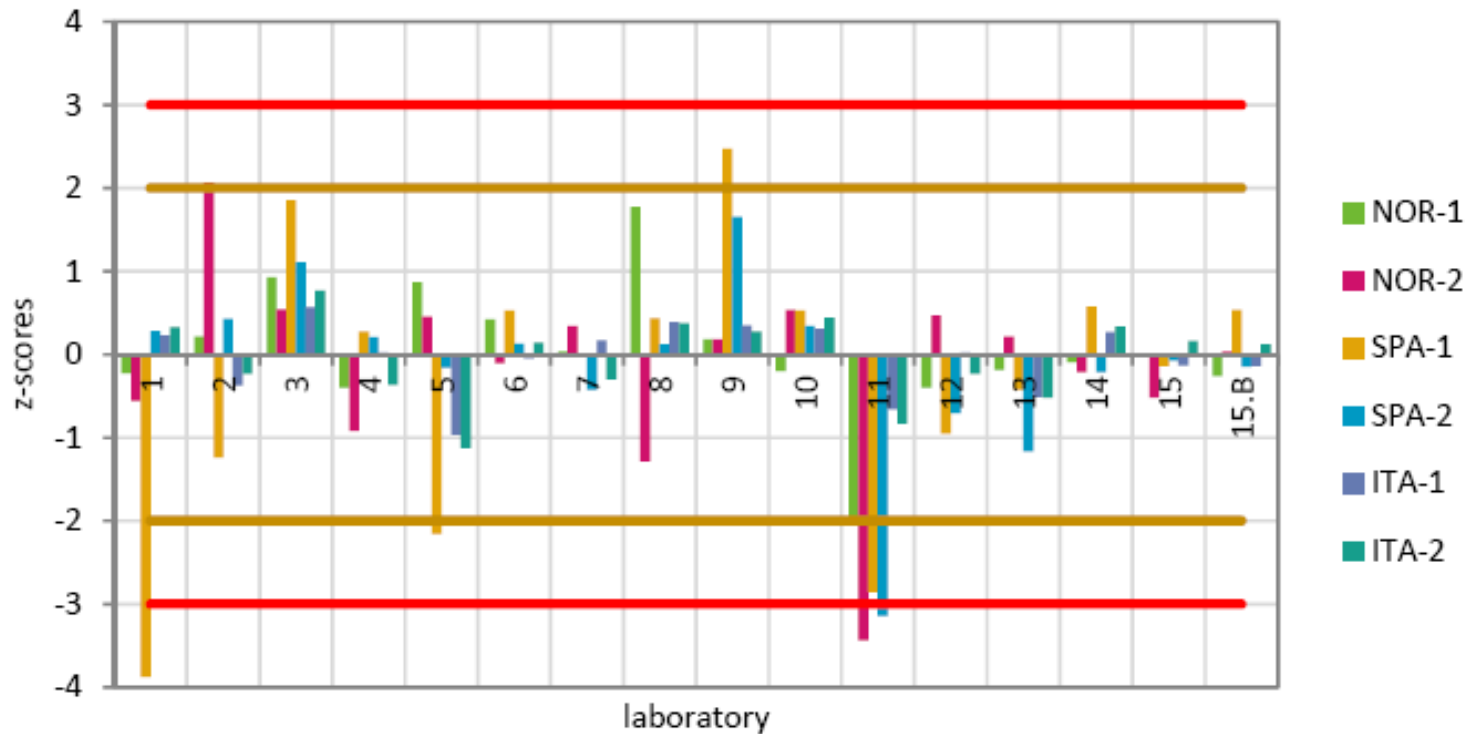
→ Evolved Gas Analysis (EGA):



Evolved Gas Analysis (EGA):



EGA methods directly **determine TC** with a reasonable precision



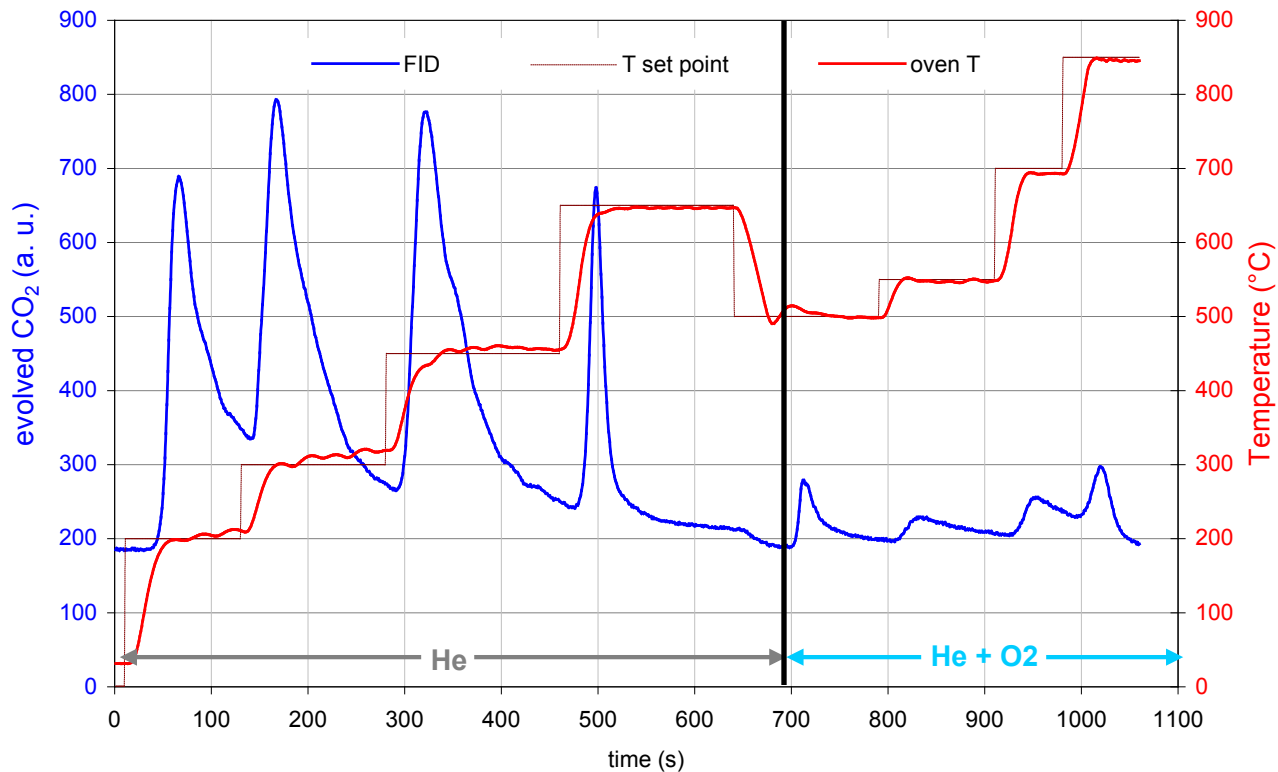
z-scores for TC calculated using a *prescribed* σ^* value (25%).

Distinguish between **EC**

directly emitted in the particulate form by combustion processes

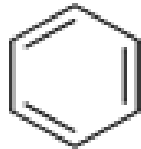
and **OC**

from both natural and anthropogenic primary and secondary sources

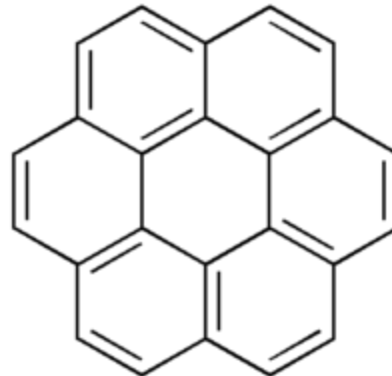


no clear theoretical **split point** between highly refractory organic molecules and “infinite” graphene structure, the basic structural element of graphite

	Volatile Organic Carbon	Particulate Organic Carbon	Elemental Carbon
H/C ratio	≈ 1	≈ 0.5	0
Volatility	high	low	0

Benzene



Coronene



Graphene

There are currently no suitable standards for atmospheric OC and EC.

But at least pure EC should be detected as 100% EC

Any organic molecule (or mixture) should be detected as 100% OC

IMPROVE (up to 550° C): all OC does not evolve during step 1

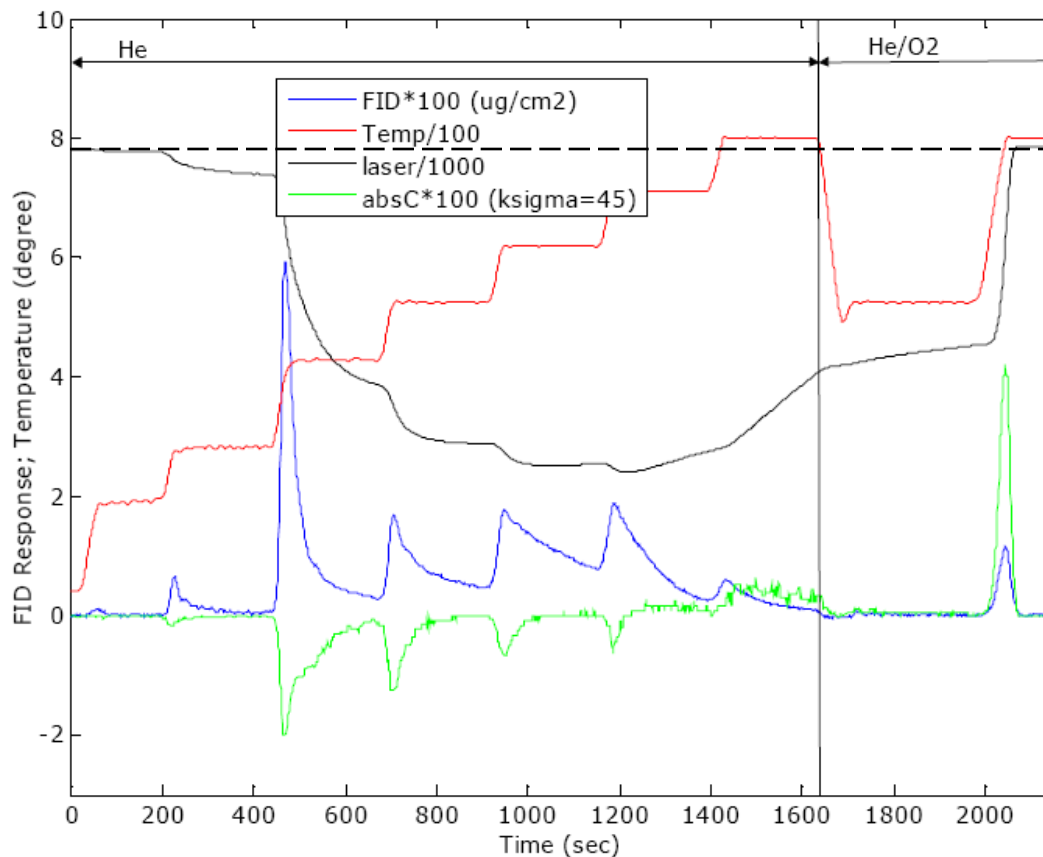
NIOSH (up to 850° C): a fraction of EC can be combusted during the step 1

EUSAAR_2 (up to 650° C): best compromise

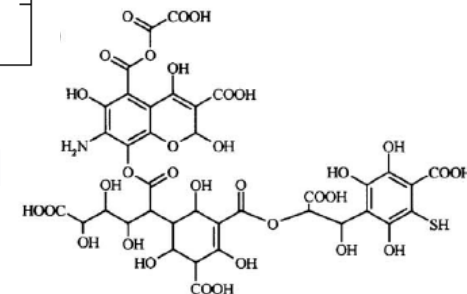
max 5 % of EC evolves in He

min 80 % of OC evolves in He

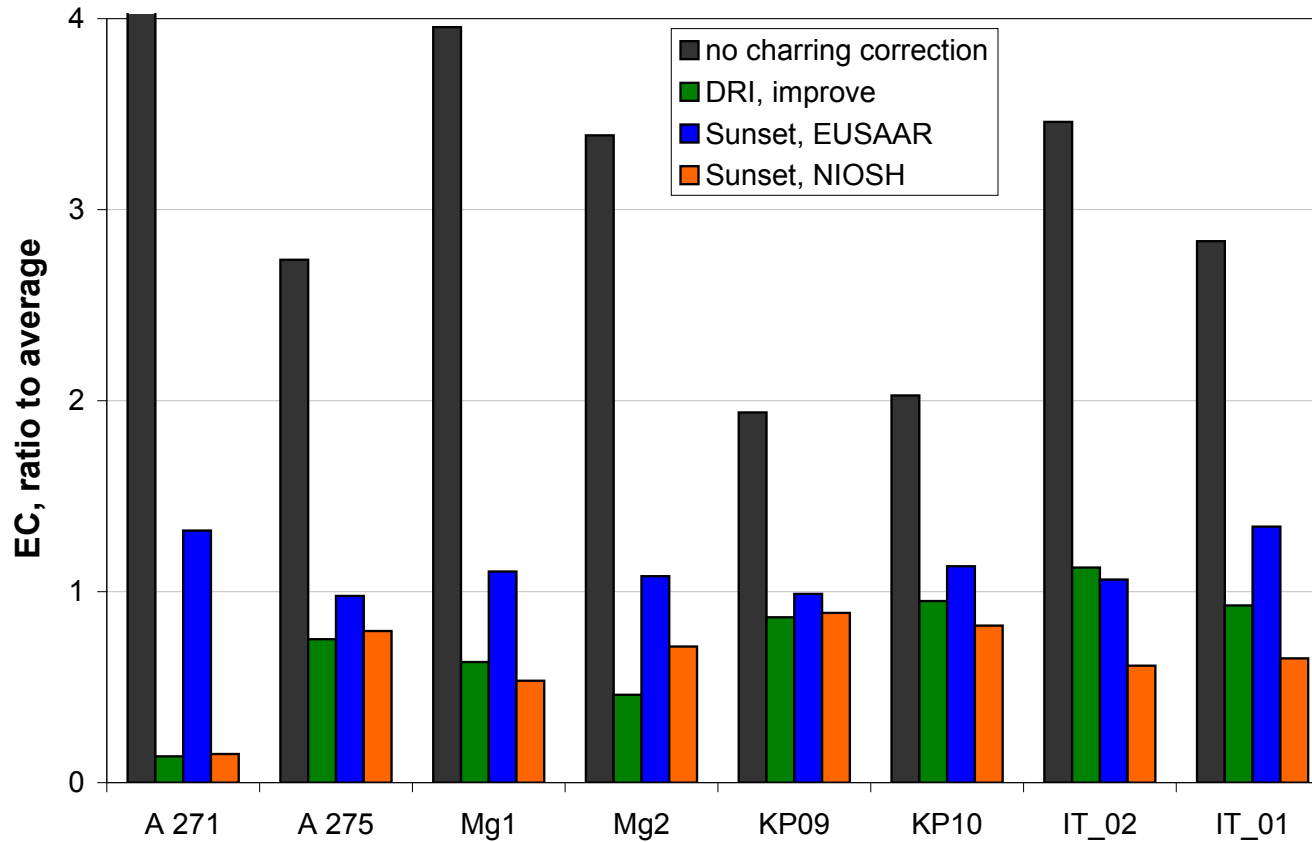
Pyrolysis produces EC-like species during the analysis (Charring):



Thermal – optical analysis of fulvic acid



Not correcting for charring => error in EC determination larger than 400%.



EC determined with 4 different thermal methods
in 8 samples collected at various sites in Europe

Charring must be corrected.

The optical correction of charring assumes that:

PC evolves from the filter before EC.

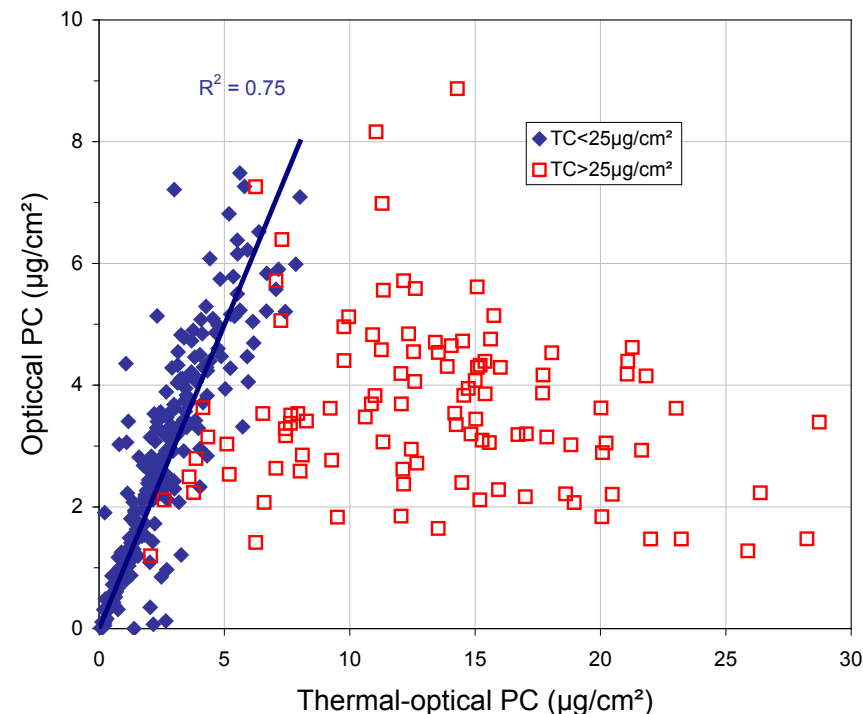
PC has the same specific light absorption cross section as EC.

which is not always true

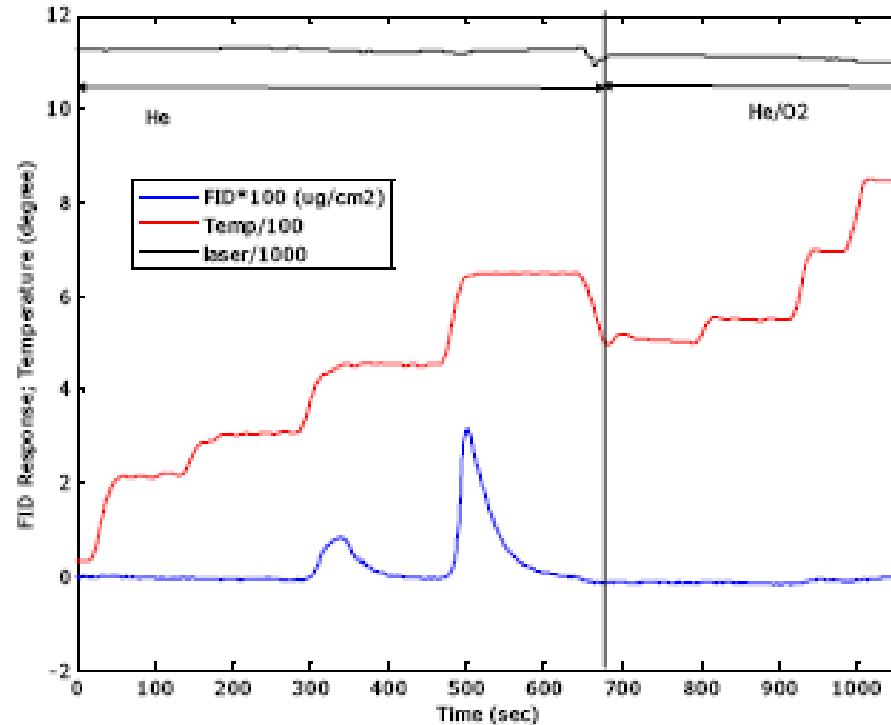
⇒ **Need for charring limitation**

With longer steps at low temperature,
EUSAAR_2 limits charring

However,
**charring correction can be inaccurate
for loadings > 25 $\mu\text{gC} / \text{cm}^2$**



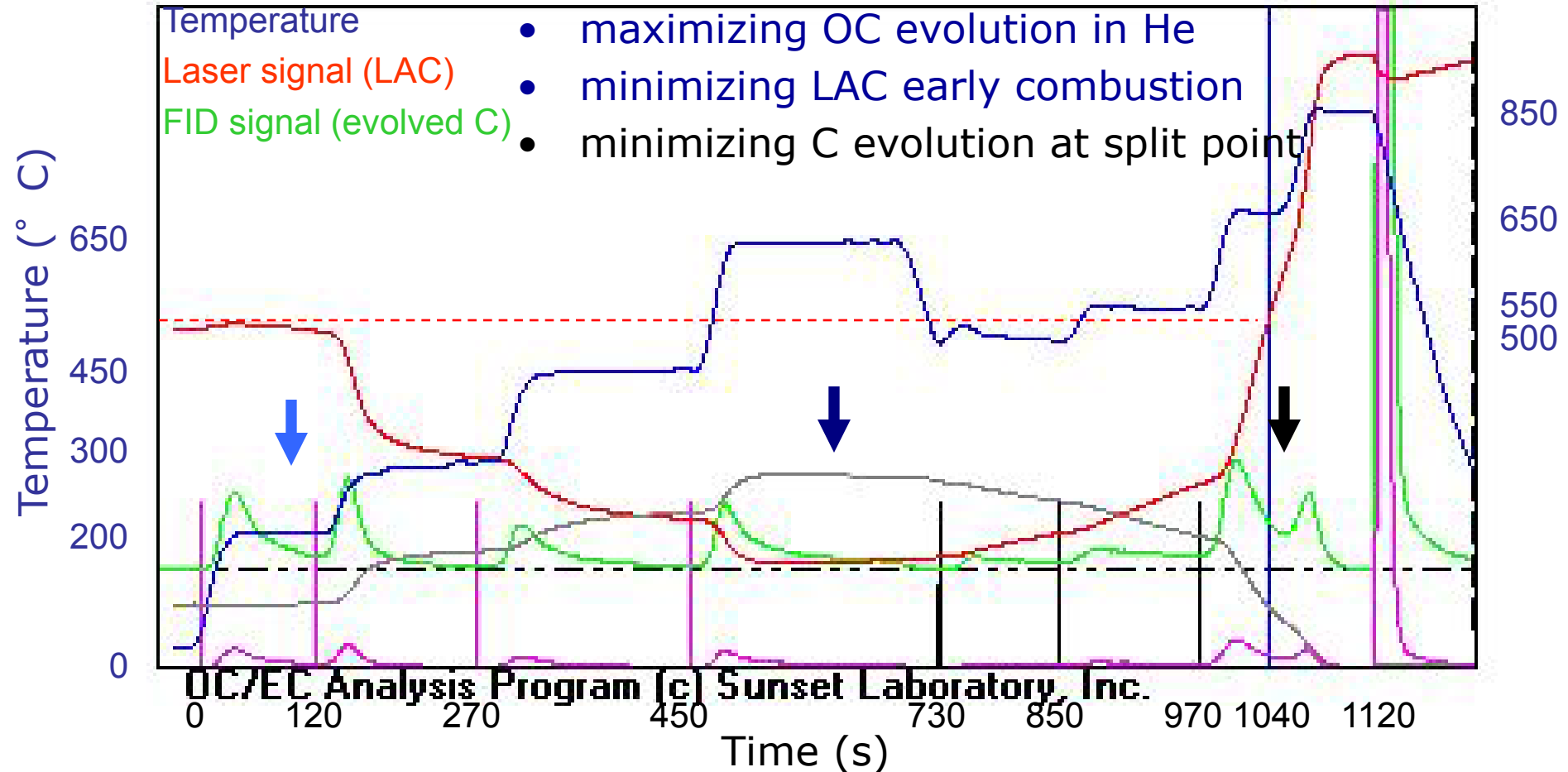
Carbonate can interfere with the determination of OC and/or EC.



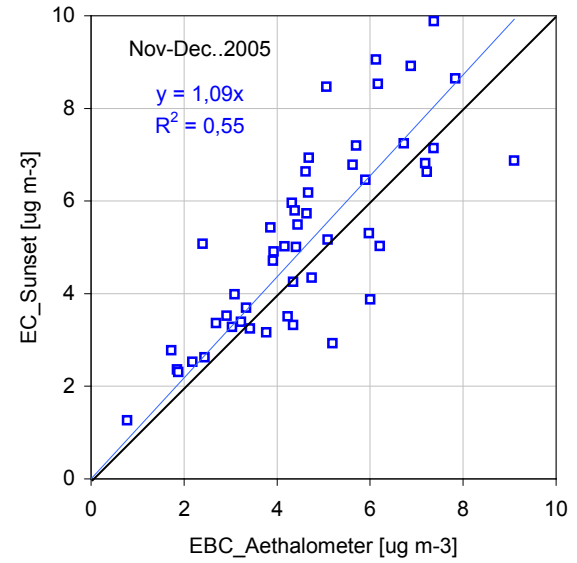
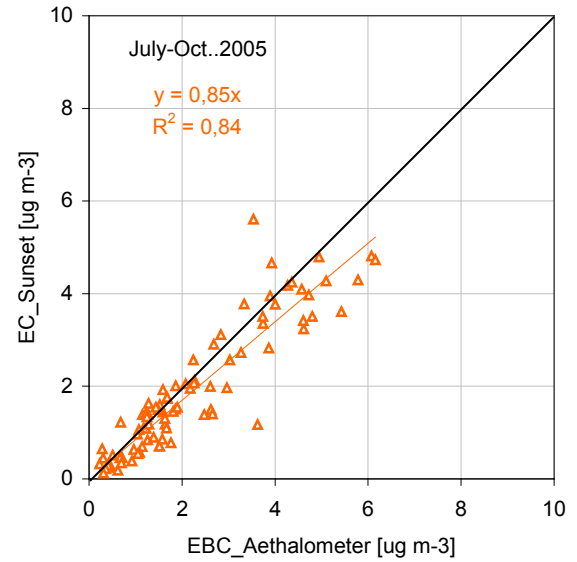
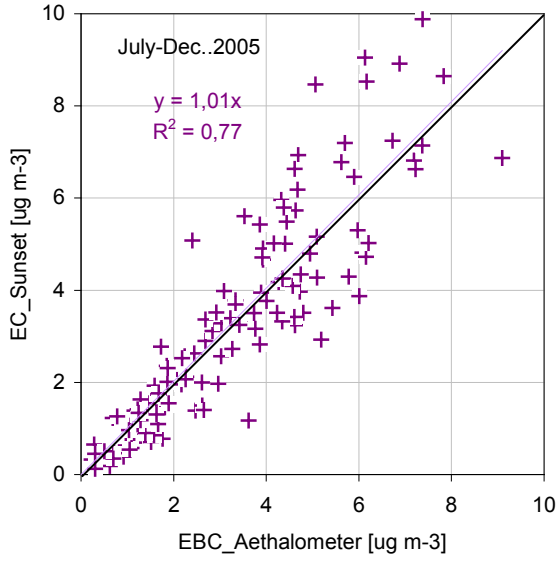
**C from natural calcite evolves in the OC4 fraction with EUSAAR_2
⇒ No interference with EC**

Protocol EUSAAR_2

- minimizing charring
- maximizing OC evolution in He
- minimizing LAC early combustion
- minimizing C evolution at split point



Comparison between Elemental Carbon and Equivalent Black Carbon



Conclusions:

The thermal-optical protocol EUSAAR_2 overcomes a series of challenges related to the measurement of “soot” in PM:

1. Robust determination of TC
2. Sound speciation of TC
3. Minimized biases in OC and EC determination
4. Enhanced precision and “related” to EBC

Towards a reference method:

1. The thermal – optical method applying the EUSAAR_2 protocol is the **standard for the EMEP network**
2. The thermal – optical method (protocol t.b.d.) will be the **EN standard method**

The question of sampling artifacts remains more critical.

Sampling artifacts affect the determination of PM mass and atmospheric particulate OC concentrations.

But sampling artifacts are not addressed by the EN standard EN12341:2014 for measuring PM mass concentrations.



Danke !

Additional Material

EUSAAR-2 protocol

Carrier gas	Temp °C	Time s
OC1_He	200	120
OC2_He	300	150
OC3_He	450	180
OC4_He	650	180
EC1_He/O ₂	500	120
EC2_He/O ₂	550	120
EC3_He/O ₂	700	70
EC4_He/O ₂	850	80

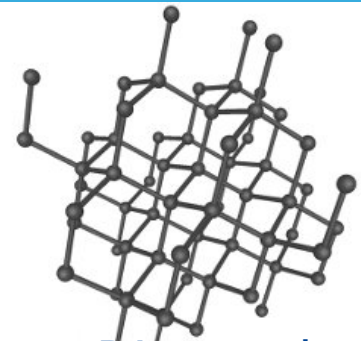
Sensitivity of the EC value to the position of the split point

The precision of the laser signal measurement translates into uncertainties in EC determination.

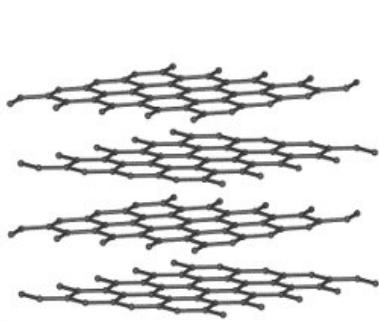
The temperature protocol should be such that the slope of the carbon peak evolving at the split point is as small as possible.

EUSAAR_2

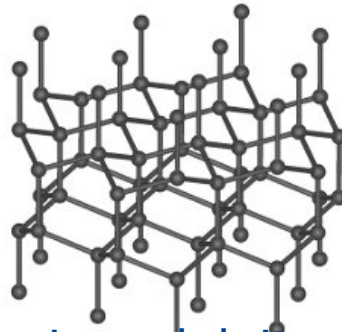
$\pm 3\%$ uncertainty in laser signal $\Rightarrow \pm 10\%$ uncertainty in EC



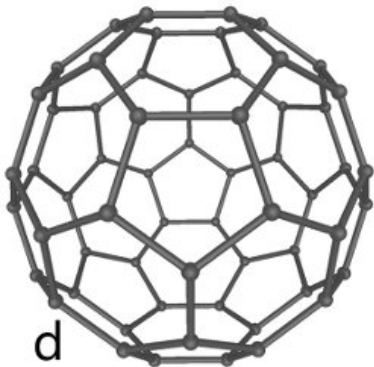
a Diamond



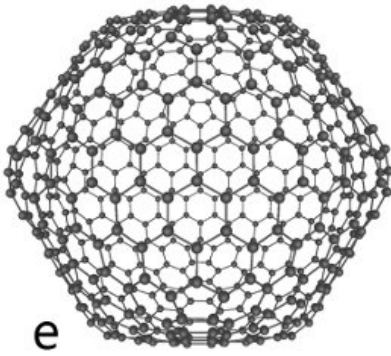
b Graphite



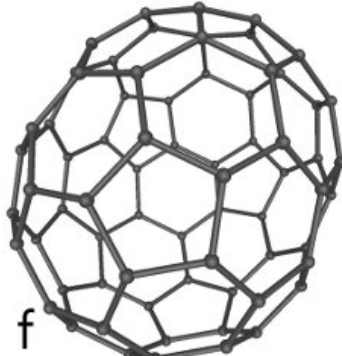
c Lonsdaleite



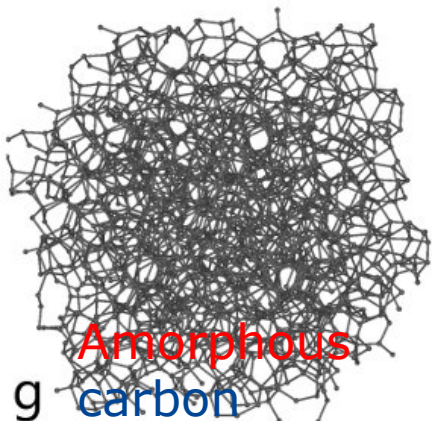
d C60



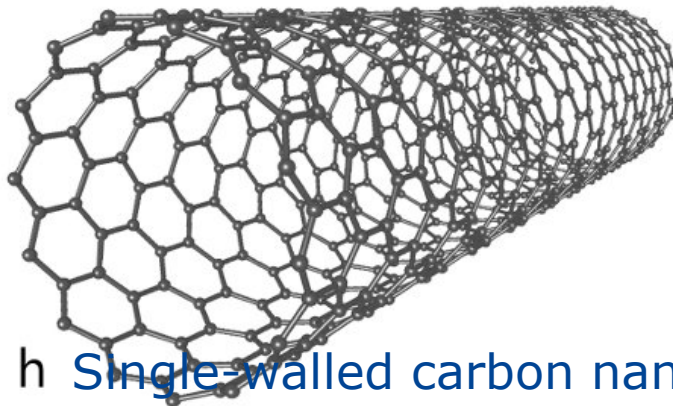
e C540



f C70 Fullerenes



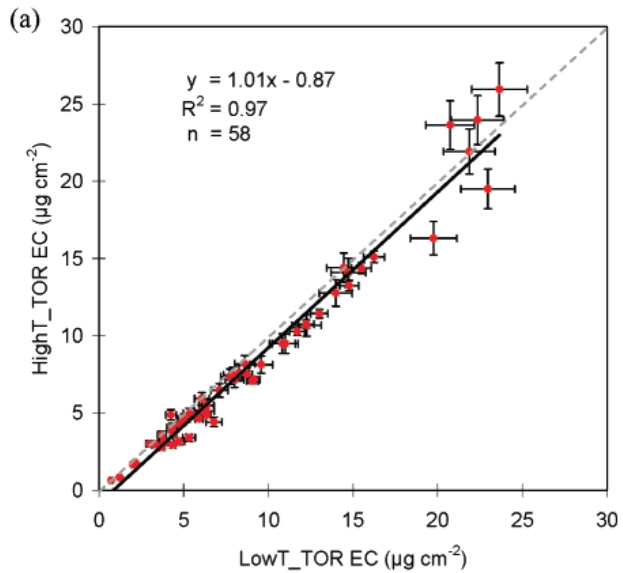
g Amorphous carbon



h Single-walled carbon nanotube

A wonderful image released by Michael Ströck under the GNU Free Documentation License:

The structures of eight allotropes of carbon



Comparisons between determined in samples from Fresno with the LowT and HighT protocols by (a) TOR, and (b) TOT (Chow et al., 2004).

